

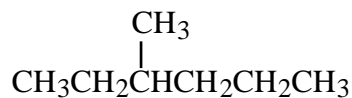
Name: \_\_\_\_\_

Period: \_\_\_\_

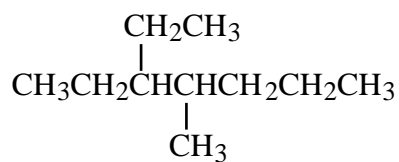
### I. Alkanes

(1) Name the following alkanes.

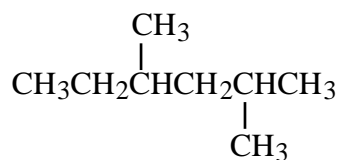
(a)



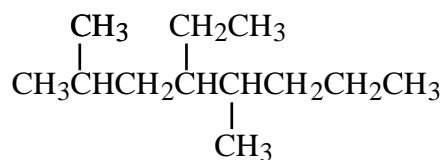
(b)



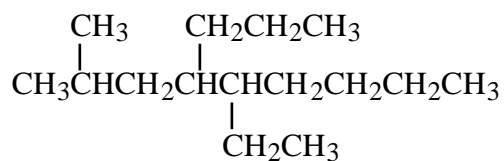
(c)



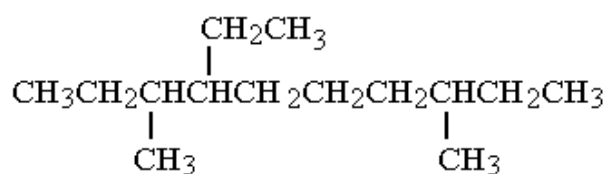
(d)



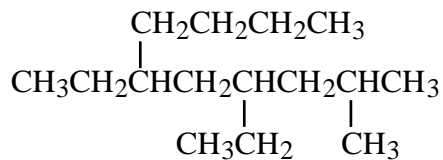
(e)



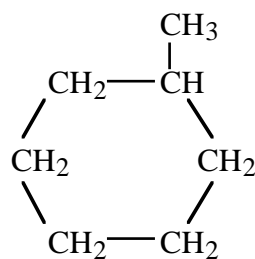
(f)



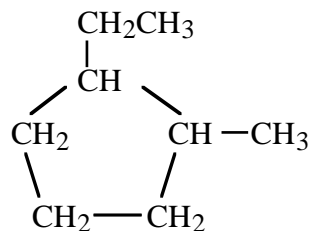
(g)



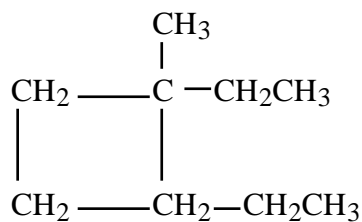
(h)



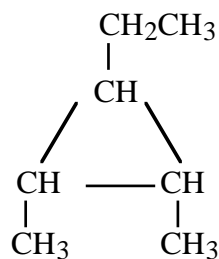
(i)



(j)



(k)



(2) Draw the following alkanes.

(a) 2-Methylpropane

(g) 3-Ethyl-2-methyl-5,6-dipropylnonane

(b) 3-Ethyl-2-methylhexane

(h) 1-Ethylcyclopropane

(c) 2,2-Dimethylbutane

(i) 1-Propyl-2,3-dimethylcyclobutane

(d) 3-Ethyl-2,5-dimethylheptane

(j) 1,2-Diethyl-3-methylcyclopentane

(e) 3,3,5-Trimethyloctane

(k) 1,3-Dimethylcyclohexane

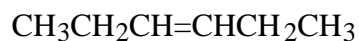
(f) 5-Butyl-2,3,6-trimethyldecane

(3) Name and draw the five structural isomers of  $C_6H_{14}$ .

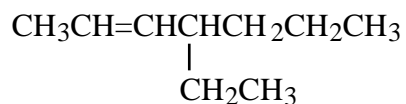
## II. Alkenes

(1) Name the following alkenes.

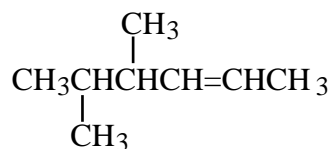
(a)



(b)



(c)



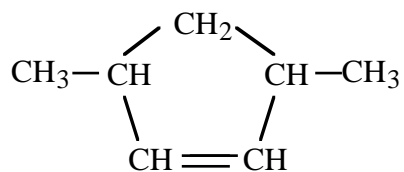
(2) Draw the following alkenes.

(a) 2-Methylpropene

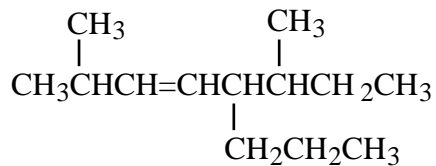
(b) 3-Ethyl-4-methyl-2-pentene

(c) 3-Ethyl-2,4-dimethyl-3-heptene

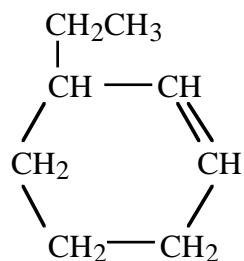
(d)



(e)



(f)

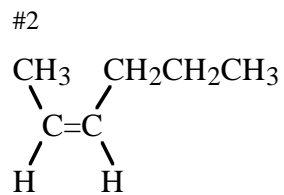
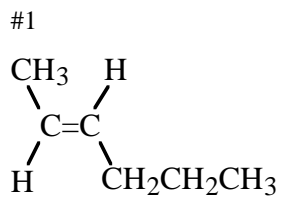


(d) 1-methylcyclopentene

(e) 6-Ethyl-3,3-dimethyl-4-nonene

(f) 3-Ethyl-4,5-dimethylcyclohexene

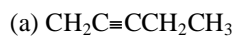
(3) (a) Name the two stereoisomers of 2-hexene.



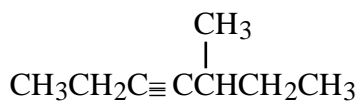
(b) Name and draw the two stereoisomers of 3-hexene.

### III. Alkynes

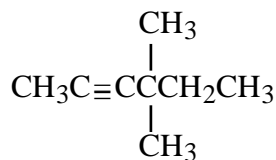
(1) Name the following alkynes.



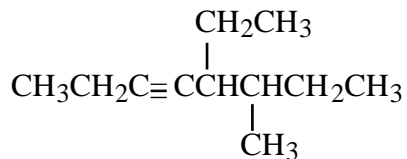
(b)



(c)



(d)



(2) Draw the following alkynes.

(a) 4-Ethyl-2-heptyne

(c) 2,5,6-Trimethyl-3-octyne

(b) 3-Ethyl-5-methyl-1-hexyne

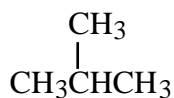
(d) 3,8-Dimethyl-5-decyne

Answers:

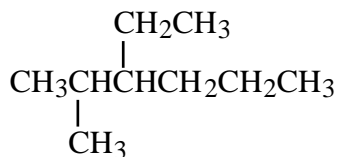
### I. Alkanes

- (1) (a) 3-Methylhexane  
(b) 3-Ethyl-4-methylheptane  
(c) 2,4-Dimethylhexane  
(d) 4-Ethyl-2,5-dimethyloctane  
(e) 5-Ethyl-2-methyl-4-propylnonane  
(f) 4-Ethyl-3,8-dimethyldecane  
(g) 4,6-Diethyl-2-methyldecane  
(h) 1-Methylcyclohexane  
(i) 1-Ethyl-2-methylcyclopentane  
(j) 1,2-Diethyl-1-methylcyclobutane  
(k) 1-Ethyl-2,3-dimethylcyclopropane

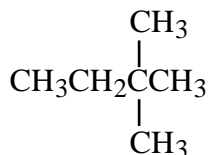
(2) (a)



(b)

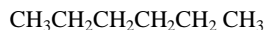


(c)

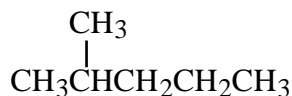


(3)

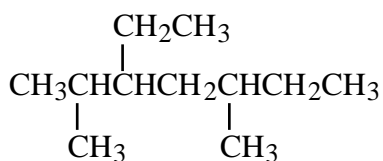
1. Hexane



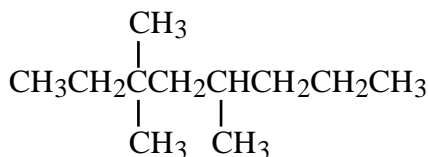
2. 2-Methylpentane



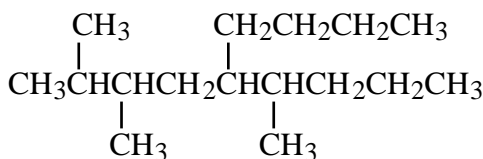
(d)



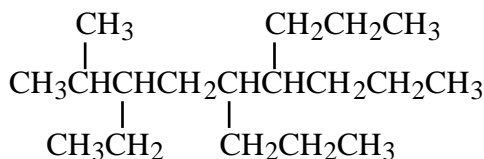
(e)



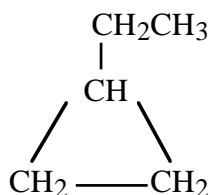
(f)



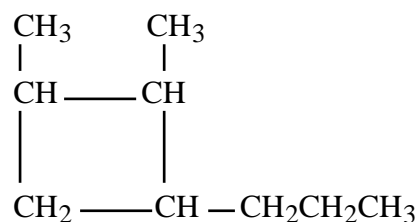
(g)



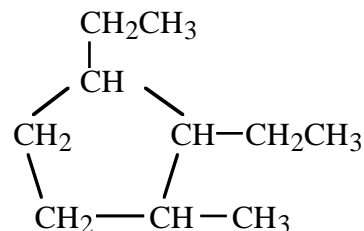
(h)



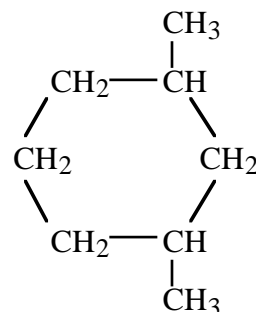
(i)



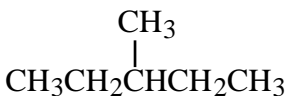
(j)



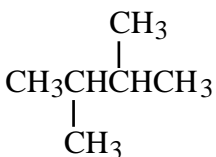
(k)



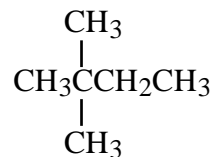
3. 3-Methylpentane



4. 2,3-Dimethylbutane



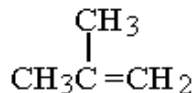
5. 2,2-Dimethylbutane



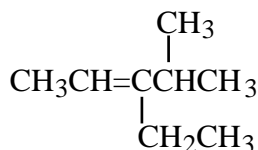
## II. Alkenes

- (1) (a) 3-hexene  
 (b) 4-Ethyl-2-heptene  
 (c) 4,5-Dimethyl-2-hexene  
 (d) 3,5-Dimethylcyclopentene  
 (e) 2,6-Dimethyl-5-propyl-3-octene  
 (f) 3-Ethylcyclohexene

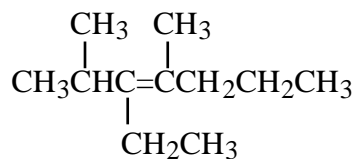
(2)(a)



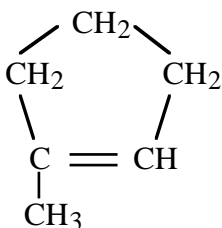
(b)



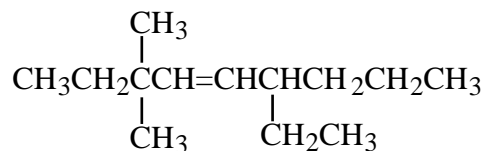
(c)



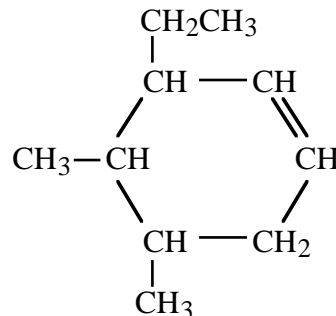
(d)



(e)



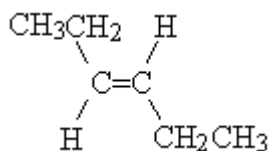
(f)



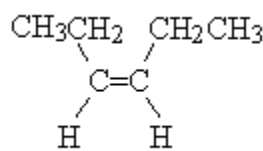
(3) (a) #1: trans-2-hexene (or (E)-2-hexene)

#2: cis-2-hexene (or (Z)-2-hexene)

(b) trans-3-hexene (or (E)-3-hexene)



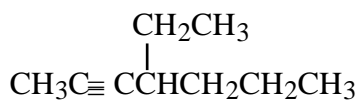
cis-3-hexene (or (Z)-3-hexene)



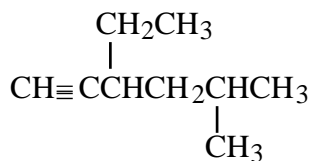
## III. Alkynes

- (1) (a) 2-pentyne  
 (b) 5-Methyl-3-heptyne  
 (c) 4,4-Dimethyl-2-hexyne  
 (d) 5-Ethyl-6-methyl-3-octyne

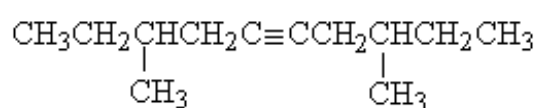
(a)



(b)



(d)



(c)

